

## Isolation of the New (-)-(3*R*,4*S*)-4-Hydroxymellein from the Fungus *Septoria nodorum* Berk

Michel Devys, Michel Barbier

Institut de Chimie des Substances Naturelles, CNRS,  
Avenue de la Terrasse, 91198 Gif-sur-Yvette Cedex,  
France

Jean-François Bousquet, Albert Kollmann

Station de Pathologie Végétale, CNRA,  
78026 Versailles Cedex, France

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The new (-)-(3*R*,4*S*)-4-hydroxymellein (**6**) is isolated from the fungus *Septoria* (*Phaeosphaeria*) *nodorum* Berk, together with the previously reported isomer **5**. The results so far obtained in this series of metabolites are discussed in relationship with biosynthetic considerations.

## Introduction

The fungus *Septoria* (*Phaeosphaeria*) *nodorum* Berk is a common parasite of wheat. It is responsible for a fungal disease which seriously affects crops (septorioses). This fungus produces in cultures a series of highly potent phytotoxins which were previously investigated. Thus, we reported on the isolation of (-)-(3*R*)-mellein (ochracin) (**4**) [1], (3*R*)-O-methylmellein, (-)-(3*R*,4*R*)-4-hydroxymellein (**5**), (-)-(3*R*)-7-hydroxymellein, mycophenolic acid [2], septorine [3], N-methoxyseptorine [4], and N-methoxyseptorinol [5]. Undoubtedly, the observed phytotoxic activity of this fungus is due to the summation of these various metabolites, some of them being endowed of a particularly strong biological activity. Mellein was shown to produce an important reduction in CO<sub>2</sub> net assimilation and an increase in stomatal resistance in seedling leaves of wheat [6–10].

In the present publication, the isolation of the new isomer (-)-(3*R*,4*S*) (**6**) from the culture medium of this fungus is reported. The results so far observed are discussed in relationship with the fact

that all the isocoumarins up to now isolated from *Septoria nodorum* are derived from (-)-mellein (**4**).

## Results and Discussion

The culture medium of the fungus *Septoria nodorum* Berk was extracted by ethyl acetate according to a previously [11] reported method. The concentrated extract was submitted to HPLC fractionation (UV detection), leading to the known (-)-(3*R*,4*R*)-4-hydroxymellein (**5**) [2], plus the new compound (-)-(3*S*,4*R*)-4-hydroxymellein (**6**).

The structures of these substances **5** and **6** were determined on the basis of physicochemical data, by comparing with previously reported results (m.p. and  $(\alpha)_D$  as represented in Fig. 1). In this series, the coupling constants of the protons at C-3 and C-4 are determinant for the attribution of the stereochemistry. As the absolute configurations were established for (-)-mellein (**1**) and (+)-(3*S*,4*S*)-4-hydroxymellein (**2**) [12, 14], it was possible to fix the configurations of the products isolated from *Septoria nodorum*. The stereochemistry of the new product **6** was confirmed by irradiation at the signal of the methyl group at 1.55 ppm, resulting in two coalescent signals (2H) for the two vicinal protons at C-3 and C-4 (one peak). Hence, it is of course deduced that the coupling constant is not noticeable between these two protons. The dihedral angle determined from a molecular model gives a value of 70° which corresponds to a coupling constant of ca. 0.3 Hz [16]. By now, three of the four possible isomers of 4-hydroxymellein are known [2, 12, 14, 15]. To our knowledge, the (+)-isomer (3*S*,4*R*) **3** still remains to be found and could exist in strains producing the corresponding (+)-mellein [14]. A diacetate of **6** was prepared which exhibited MS and <sup>1</sup>H NMR spectra in agreement with the proposed stereochemistry. In particular, the vicinal <sup>3</sup>J coupling constant observed in the <sup>1</sup>H NMR of this diacetate is of approximately 1 Hz. It appears from the molecular models that two conformers are possible for substance **6**, each requiring some energy for transformation. The dihedral angle 3-C–H 4-C–H is of ca. 70° when the methyl group is  $\beta$ -axial which leads to the observed coupling constant according to the Karplus equation [16]. But in the conformer having this methyl group  $\beta$ -equatorial, this angle is

Reprint requests to Dr. M. Barbier.

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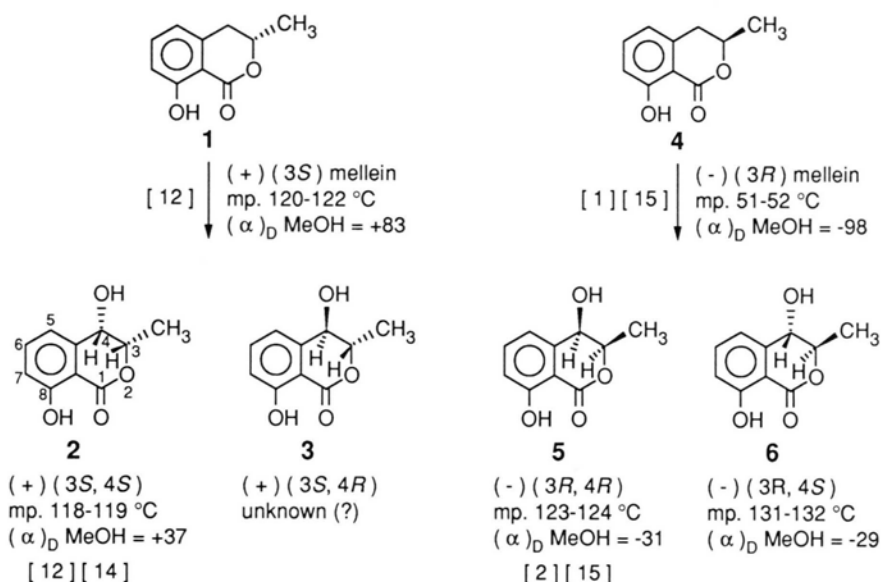


Fig. 1. Absolute configurations of the two melleins (ochracins) and of the four possible 4-hydroxymelleins.

of 180°, a value which must give a higher  $J$  (between 6 and 15 Hz). Hence, two conclusions may be drawn: 1) the compound **6** isolated from *Septoria nodorum* has the 3- $\beta$ -axial conformation, 2) the possibility that two conformers of the *cis*-series **2** and **5** must be small (or negligible), whereas such differences should be much bigger in the *trans*-series **3** and **6**.

The presence of the two isomers **5** and **6** in the culture medium of *Septoria nodorum* is in agreement with the demonstration that the biosynthesis of (+)-(3*S*,4*S*)-4-hydroxymellein (**2**) proceeds through oxidation of (+)-mellein (**1**) by molecular oxygen [14]. The previously found (-)-O-methylmellein and (-)-7-hydroxymellein, together present in *Septoria nodorum* with (-)-mellein [2] and the (-)-4-hydroxymelleins, demonstrate of course the same biogenetic family.

## Experimental

Melting points have been determined with a Kofler apparatus under the microscope and are corrected. MS were carried out on an AEI MS 50 spectrometer, <sup>1</sup>H NMR on a Bruker 200 MHz apparatus (CDCl<sub>3</sub>,  $\delta$  ppm). Rotatory power was de-

termined on an electronic Perkin-Elmer polarimeter 241. TLC's on Schleicher-Schüll SiO<sub>2</sub> fluorescent films, 1 mm thickness for preparative purposes, UV observation with a Desaga lamp at 254 nm, or FeCl<sub>3</sub> spray for visualization.

The 4-hydroxymelleins **5** and **6** were extracted from the culture medium of *Septoria nodorum* according to the reported [11] method. The EtOAc concentration was submitted to HPLC (elution by ethanol-water, UV detection) and the two substances **5** and **6** recovered according to the absorption curve. SiO<sub>2</sub> TLC (CH<sub>2</sub>Cl-EtOAc 5:1), **5**  $R_f$  0.60, colour reaction with FeCl<sub>3</sub>: blue, **6**  $R_f$  0.65, FeCl<sub>3</sub> pink, relative proportions 5:1. The physicochemical data of substance **5** are identical to the reported values [2] except for small differences in the m.p. and ( $\alpha$ )<sub>D</sub> due to more accurate technical facilities (such as electronic polarimeter). These values are reported on the Figure. **6**: MS  $m/z$  (%), 194  $M^+$  (100), 150 (70), 121 (80), 122 (80); <sup>1</sup>H NMR: 1.55 (d, 3H, CH<sub>3</sub>), 4.50 (enlarged signal, 2H, H-3 and H-4), giving a singlet by irradiation of the methyl signal at 1.55; aromatic protons: 7.04 (d, 1H, H-5,  $J$  = 8 Hz), 7.55 (dd, 1H, H-6,  $J$  = 8 Hz), 6.95 (d, 1H, H-7,  $J$  = 8 Hz), 11 (s, 1H, 8-OH). The diacetate of **6** was prepared by treatment with acetic anhydride in the presence of anhydrous pyridine (10 mg **6**, 6 drops of each reagent, 1 h at 37 °C, drying under *in vacuo* overnight at room tempera-

ture). SiO<sub>2</sub> TLC in pentane–ethyl acetate 1:1, *R<sub>f</sub>* 0.70, MS *m/z* (%) 278 M<sup>+</sup> (2), 236 M–42<sup>+</sup> (100), 194 M–42–42<sup>+</sup> (45), 176 M–42–60 (100), 150 (100), 149 (75), 43 CH<sub>3</sub>CO<sup>+</sup> (95). <sup>1</sup>H NMR (CDCl<sub>3</sub>): 1.35 (d, 3H, CH<sub>3</sub>), 2.13 and 2.40 (s, 3H each, CH<sub>3</sub>COO), 4.80 (m, 1H, H-3), 5.83 (d, 1H, H-4, *J* = 1 Hz), 7.18 (d, 1H, H-5, *J* = 8 Hz), 7.36

(d, 1H, H-7, *J* = 8 Hz), 7.66 (dd, 1H, H-6, *J* = 8 Hz).

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